

# **Safety Data Sheet** 2% Chlorine/ 98% Nitrogen

# **Section 1: Product and Company Identification**

Purity Cylinder Gases,Inc.

2580 28th St SW Wyoming, MI 49519 P: (616)532-2375 www.puritygas.com

Product Code: 2% Chlorine/ 98% Nitrogen

Synonyms: Recommended Use: **Usage Restrictions:** 

## **Section 2: Hazards Identification**



#### **Hazard Classification:**

Acute Aquatic Toxicity (Category 1) Eye Effects (Category 2.A) Gases Under Pressure Specific target organ toxicity (Single Exposure) (Category 3)

#### **Hazard Statements:**

Causes serious eve irritation Contains gas under pressure; may explode if heated May cause respiratory irritation; Very toxic to aquatic life

### **Precautionary Statements**

#### Prevention:

Avoid breathing dust/fume/gas/mist/ vapors/spray. Wash thoroughly after handling. Wear eye protection/face protection. Use only outdoors or in a well-ventilated area. [In case of inadequate ventilation] wear respiratory protection.

#### Response:

If in eyes: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.

Call a poison center or doctor if you feel unwell.

If inhaled: Remove person to fresh air and keep comfortable for breathing.

#### Storage:

Store locked up.

Purity Cylinder Gases,Inc. Generated by the SDS Manager from AsteRisk, LLC. All Rights Reserved Date of Preparation: 07/23/2021 15:51:06 Protect from sunlight.

Store in a well-ventilated place. Keep container tightly closed.

#### Disposal:

Dispose of contents and/or container in accordance with applicable regulations.

# Section 3: Composition/Information on Ingredients

	CAS#	Concentration
Chlorine	7782-50-5	2
Nitrogen	7727-37-9	98

	Chemical Substance	Chemical Family	Trade Names
Chlorine	CHLORINE	Halogens	CHLORINE MOLECULAR; DIATOMIC CHLORINE; DICHLORINE; MOLECULAR CHLORINE; UN 1017; CI2
Nitrogen	NITROGEN, COMPRESSED GAS	Inorganic gases	DIATOMIC NITROGEN; DINITROGEN; NITROGEN; NITROGEN-14; NITROGEN GAS; UN 1066; N2

## **Section 4: First Aid Measures**

	Skin Contact	Eye Contact	Ingestion	Inhalation	Note to Physicians
Chlorine	Wash skin with soap and water for at least 15 minutes while removing contaminated clothing and shoes. Get immediate medical attention. Thoroughly clean and dry contaminated clothing before reuse. Destroy contaminated shoes.	Immediately flush eyes with plenty of water for at least 15 minutes. Then get immediate medical attention.	Not likely route of exposure.	If adverse effects occur, remove to uncontaminated area. Give artificial respiration if not breathing. If breathing is difficult, oxygen should be administered by qualified personnel. Get immediate medical attention.	For inhalation, consider oxygen. Avoid gastric lavage or emesis.
Nitrogen Wash exposed skin with soap and Flush eyes		Flush eyes with plenty of water.	If a large amount is swallowed, get medical attention.	If adverse effects occur, remove to uncontaminated area. Give artificial respiration if not breathing. If breathing is difficult, oxygen should be administered by qualified personnel. Get immediate medical attention.	For inhalation, consider oxygen.

# **Section 5: Fire Fighting Measures**

	Suitable Extinguishing Media	Products of Combustion	Protection of Firefighters		
Chlorine	Non-flammable. Use appropriate extinguishing media for surrounding fire.	Non-flammable	<ul> <li>Full-body encapsulating chemical protective suit with positive pressure self-contained breathing apparatus</li> <li>Non-flammable.</li> </ul>		
Nitrogen	Non-flammable. Use suitable extinguishing media for surrounding fire. Cylinders may rupture or explode if exposed to heat.	Non-flammable	<ul> <li>Respiratory protection may be needed for frequent or heavy exposure.</li> </ul>		

Purity Cylinder Gases,Inc. page 2 of 6

## **Section 6: Accidental Release Measures**

	Personal Precautions	Environmental Precautions	Methods for Containment
Chlorine	Keep unnecessary people away, isolate hazard area and deny entry. Ventilate closed spaces before entering. Evacuate area and downwind locations.	Avoid contact with combustible materials.	Stop leak if possible without personal risk. Reduce vapors with water spray. Dig holding area such as lagoon, pond or pit for containment. Trap spilled material at bottom in deep water pockets, excavated holding areas or within sand bag barriers.
Nitrogen	Keep unnecessary people away, isolate hazard area and deny entry. Stay upwind and keep out of low areas.	No significant effects from contamination expected.	Stop leak if possible without personal risk.

	Methods for Cleanup	Other Information
Chlorine	Collect runoff for disposal as potential hazardous waste. Dike for later disposal. Absorb with sand or other non-combustible material. Add an alkaline material (lime, crushed limestone, sodium bicarbonate, or soda ash). Add an alkaline material (lime, crushed limestone, sodium bicarbonate, or soda ash). Absorb with activated carbon. Collect spilled material using mechanical equipment.	Notify Local Emergency Planning Committee and State Emergency Response Commission for release greater than or equal to RQ (U.S. SARA Section 304). If release occurs in the U.S. and is reportable under CERCLA Section 103, notify the National Response Center at (800)424-8802 (USA) or (202)426-2675 (USA).
Nitrogen	N/A	N/A

## **Section 7: Handling and Storage**

	Handling	Storage
Chlorine	Notify State Emergency Response Commission for storage or use at amounts greater than or equal to the TPQ (U.S. EPA SARA Section 302). SARA Section 303 requires facilities storing a material with a TPQ to	Store and handle in accordance with all current regulations and standards. Protect from physical damage. Keep separated from incompatible
Nitrogen	participate in local emergency response planning (U.S. EPA 40 CFR 355.30).  Store and handle in accordance with all current regulations and standards.	substances. Store outside or in a detached building.  Keep separated from incompatible substances.
I viii ogen	Subject to storage regulations: U.S. OSHA 29 CFR 1910.101.	recep separated from moornpatible substances.

## **Section 8: Exposure Controls/Personal Protection**

	Exposure Guidelines
Chlorine	CHLORINE: 1 ppm (3 mg/m3) OSHA ceiling 0.5 ppm (1.5 mg/m3) OSHA TWA (vacated by 58 FR 35338, June 30, 1993) 1 ppm (3 mg/m3) OSHA STEL (vacated by 58 FR 35338, June 30, 1993) 0.5 ppm ACGIH TWA 1 ppm ACGIH STEL 0.5 ppm (1.45 mg/m3) NIOSH recommended ceiling 15 minute(s)
Nitrogen	NITROGEN, COMPRESSED GAS: NITROGEN: ACGIH (simple asphyxiant)

### **Engineering Controls**

Handle only in fully enclosed systems.

	Eye Protection	Skin Protection	Respiratory Protection
Chlorine	Wear splash resistant safety goggles with a face shield. Provide an emergency eye wash fountain and quick drench shower in the immediate work area.	Wear appropriate chemical resistant clothing.	Full-body encapsulating chemical protective suit with positive pressure self-contained breathing apparatus
Nitrogen	Eye protection not required, but recommended.	Protective clothing is not required.	Respiratory protection may be needed for frequent or heavy exposure.

### **General Hygiene considerations**

- Avoid breathing vapor or mist
- Avoid contact with eyes and skin
- Wash thoroughly after handling and before eating or drinking

## **Section 9: Physical and Chemical Properties**

Physical	Appearance	Color	Change in	Physical	Odor	Taste
State			Appearance	Form		

Purity Cylinder Gases,Inc. page 3 of 6 Date of Preparation: 07/23/2021 15:51:06

	Physical State	Appearance	Color	Change in Appearance	Physical Form	Odor	Taste
Chlorine	Gas	Yellow or green	Yellow or green	N/A	Gas	Distinct odor, irritating odor	N/A
Nitrogen	Gas	Clear	Colorless	N/A	Gas	Odorless	Tasteless

	Flash Point		Partition Coefficient	Autoignition Temperature	Upper Explosive Limits	Lower Explosive Limits
Chlorine	Not combustible (does not burn). However, chlorine is a strong oxidizing agent and is a serious fire risk.	Not available	Not available	Not available	Not available	Not available
Nitrogen	Not flammable	Not available	Not available	Nonflammable	Nonflammable	Nonflammable

	Boiling Point	Freezing Point	Vapor Pressure	Vapor Density	Specific Gravity	Water Solubility	рH	Odor Threshold	Evaporation Rate	Viscosity
Chlorine	-29.1 F (-33.97 C)	-150 F (- 101 C)	5168 mmHg @ 21 C	2.49 (Air=1)	Not applicable	1.46% @ 0 C	Not applicable	0.01 ppm	Not applicable	0.01327 cP @ 20 C
Nitrogen	-321 F (-196 C)	-346 F (- 210 C)	760 mmHg @ -196 C	0.967 (Air=1)	Not applicable	1.6% @ 20 C	Not applicable	Not available	Not applicable	0.01787 cP @ 27 C

	Molecular Weight	Molecular Formula	Density	Weight per Gallon	Volatility by Volume	Volatility	Solvent Solubility
Chlorine	70.906	Cl2	3.214 g/L @ 0 C	Not available	100%	Not applicable	Soluble: Alkali
Nitrogen	28.0134	N2	1.2506 g/L	Not available	100%	1	Soluble: Liquid ammonia

# Section 10: Stability and Reactivity

	Stability	Conditions to Avoid	Incompatible Materials
Chlorine	Stable at normal temperatures and pressure.	Stable at normal temperatures and pressure.	Combustible materials, bases,
	It reacts with water to form a weak, highly	It reacts with water to form a weak, highly	metals, halogens, metal salts,
	corrosive solutions of hydrochloric acid and	corrosive solutions of hydrochloric acid and	reducing agents, amines, metal
	hypochlorous acid, which can decompose to	hypochlorous acid, which can decompose to	carbide, metal oxides, oxidizing
	hydrochloric acid and oxygen.	hydrochloric acid and oxygen.	materials, halo carbons, acids
Nitrogen	Stable at normal temperatures and pressure.	Stable at normal temperatures and pressure.	Metals, oxidizing materials

	Hazardous Decomposition Products	Possibility of Hazardous Reactions
Chlorine	Corrosive hydrogen chloride, hydrochloric acid and hypochlorous acid.	Will not polymerize.
Nitrogen	Oxides of nitrogen	Will not polymerize.

# **Section 11: Toxicology Information**

### **Acute Effects**

	Oral LD50	Dermal LD50	Inhalation
Chlorine	0.86 mg/L (1 hr- Rat)	Not available	Burns, chest pain, difficulty breathing, headache, dizziness, hyperactivity, emotional disturbances, bluish skin color, lung damage, death
Nitrogen	Not available	Not available	Nausea, vomiting, difficulty breathing, headache, drowsiness, dizziness, tingling sensation, loss of coordination, convulsions, coma

	Eye Irritation	Skin Irritation	Sensitization
Chlorine	Burns	Burns	Acute toxicity, Category 2, inhalation; H330: Fatal if inhaled. Skin irritation, Category 2; H315: Causes skin irritation. Eye irritation, Category 2; H319: Causes serious eye irritation. Specific Target Organ Toxicity (single exposure), Category 3; H335: May cause respiratory irritation.
Nitrogen	Contact with rapidly expanding gas may cause burns or frostbite	No information on significant adverse effects	Difficulty breathing

Purity Cylinder Gases,Inc.

page 4 of 6

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#### **Chronic Effects**

	Carcinogenicity	Mutagenicity	Reproductive Effects	Developmental Effects
Chlorine	ACGIH: A4 -Not Classifiable as a Human Carcinogen	Available.	Available.	No data
Nitrogen	Not hazardous	Not available	Not available	No data

## **Section 12: Ecological Information**

**Fate and Transport** 

	Eco toxicity	Persistence / Degradability	Bioaccumulation / Accumulation	Mobility in Environment
Chlorine	Fish toxicity: LC50 Fathead minnow: 0.07 to 0.15 (96 hour); 390 ug/L 96 hour(s) LC50 (Mortality) Orangethroat darter (Etheostoma spectabile) Invertibrate toxicity: 637.5 ug/L 1 hour(s) LC50 (Mortality) Pacific oyster (Crassostrea gigas) Algal toxicity: 50-1000 ug/L 23 hour(s) (Population) Algae,phytoplankton,algal mat (Algae) Phyto toxicity: Not available Other toxicity: 20 ug/L 96 day(s) (Growth) Watermilfoil (Myriophyllum spicatum)	The atmospheric half-life and lifetime of this material due to photolysis is estimated at 10 and 14 minutes, respectively. The half-life of free resid	Not expected	Not available
Nitrogen	Fish toxicity: Not available Invertibrate toxicity: Not available Algal toxicity: Not available Phyto toxicity: Not available Other toxicity: Not available	Not available	Not available	Not available

## **Section 13: Disposal Considerations**

Chlorii	e Subject to disposal regulations: U.S. EPA 40 CFR 262. Hazardous Waste Number(s): D001. Dispose in accordance with all applicable
	regulations.
Nitrog	n Dispose in accordance with all applicable regulations.

# **Section 14: Transportation Information**

### U.S. DOT 49 CFR 172.101

### **DOT Information For This Mixture**

Shipping Name Compressed gas, n.o.s. (Nitrogen, Chlorine)							
UN Number	UN1956						
Hazard Class	2.2						
Hazard Information	Non-Flammable Gas						

**Individual Component Information** 

	Proper Shipping Name	ID Number	Hazard Class or Division	Packing Group	Labeling Requirements	Passenger Aircraft or Railcar Quantity Limitations	Cargo Aircraft Only Quantity Limitations	Additional Shipping Description
Chlorine	Chlorine	UN1017	2.3	Not applicable	2.3; 8	Forbidden	Forbidden	Toxic-Inhalation Hazard Zone B
Nitrogen	Nitrogen, compressed	UN1066	2.2	Not applicable	2.2	75 kg or L	150 kg	N/A

**Canadian Transportation of Dangerous Goods** 

	Shipping Name	UN Number	Class	Packing Group / Risk Group
Chlorine	Chlorine	UN1017	2.3; 8	Not applicable

Date of Preparation: 07/23/2021 15:51:06

Nitrogen   Nitrogen, compressed   UN1066   2.2   Not applicable
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## **Section 15: Regulatory Information**

### **U.S. Regulations**

	CERCLA Sections	SARA 355.30	SARA 355.40
Chlorine	10 LBS RQ	100 LBS TPQ	10 LBS RQ
Nitrogen	Not regulated.	Not regulated.	Not regulated.

### **SARA 370.21**

	Acute	Chronic	Fire	Reactive	Sudden Release
Chlorine	Yes	No	No	No	Yes
Nitrogen	Yes	No	No	No	Yes

### **SARA 372.65**

Chlorine	CHLORINE	
Nitrogen	Not regulated.	

### **OSHA Process Safety**

Chlorine	1500 LBS TQ	
Nitrogen	Not regulated.	

### **State Regulations**

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	CA Proposition 65	
Chlorine	Not regulated.	
Nitrogen	Not regulated.	

### **Canadian Regulations**

	WHMIS Classification
Chlorine	A, D1A, E
Nitrogen	Α

### **National Inventory Status**

	US Inventory (TSCA)	TSCA 12b Export Notification	Canada Inventory (DSL/NDSL)
Chlorine	Listed on inventory.	Not listed.	Not determined.
Nitrogen	Listed on inventory.	Not listed.	Listed on inventory.

# **Section 16: Other Information**

	NFPA Rating
Chlorine	HEALTH=4 FIRE=0 REACTIVITY=0 SPECIAL=OX
Nitrogen	HEALTH=0 FIRE=0 REACTIVITY=0 SPECIAL=SA

0 = minimal hazard, 1 = slight hazard, 2 = moderate hazard, 3 = severe hazard, 4 = extreme hazard

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